The Structure of 4-(2,4-Dihydroxy-2,6,6-trimethylcyclohexylidene)-3-buten-2-one

By JOHN N. LOW AND PATRICK TOLLIN

Carnegie Laboratory of Physics, University of Dundee, Dundee DD1 4HN, Scotland

AND E. B. FLEISCHER

Office of the Dean, College of Arts and Sciences, University of Colorado, Boulder, Colorado, USA

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C1 C2

C21

02

C3 C4

O4

C5 C6

C61

C62

C7 C8

C9

C91

09

Abstract. $C_{13}H_{20}O_3$, $M_r = 224 \cdot 3$, orthorhombic, $Pna2_1$, $a = 25 \cdot 94$ (1), $b = 7 \cdot 804$ (5), $c = 6 \cdot 890$ (5) Å, $U = 1394 \cdot 78$ Å³, Z = 4, $D_m = 1 \cdot 03$, $D_x = 1 \cdot 07$ g cm⁻³, Cu Ka radiation, $\lambda = 1 \cdot 5418$ Å, $\mu = 5 \cdot 30$ cm⁻¹, F(000) = 488, T = 293 K, $R = 0 \cdot 069$ for 641 unique reflections. The two hydroxyl groups are *trans* to one another. The secondary hydroxyl group occupies an equatorial position. Bond lengths and angles are normal; the allene angle is 179 \cdot 5 (13)°.

Introduction. The title compound was isolated from a froth produced as a defensive mechanism by the flightless grasshopper *Romalea microptera* (Meinwald, Erickson, Hartshorn, Meinwald & Eisner, 1968). In this and a further paper (Meinwald & Hendry, 1969) it was concluded that the compound was an allenic sesquiterpene with the structural formula shown in Fig. 1. This study confirms this conclusion.

Experimental.* Refined cell dimensions measured on a Pailred diffractometer. The data were collected with Cu $K\alpha$ monochromatized radiation (graphite crystal) from a crystal mounted along the *c* axis. 1863 independent reflections measured within one quadrant, 641 unique observed reflections on merging Friedel pairs. Absorption corrections not applied. Structure solved by direct methods with *MITHRIL* (Gilmore, 1983) using *E* values obtained from *SHELX*76 (Sheldrick, 1976). Refinement (on *F*) by full-matrix least squares using *SHELX*76. H-atom positions included at calculated positions based on peak positions on a difference Fourier map. Non-hydrogen atoms refined anisotropically. H atoms included as fixed atoms with isotropic temperature factor of 0.1 Å². 144



Fig. 1. View of molecule showing conformation and atomic numbering.

Table 1. Coordinates $(\times 10^4)$ for non-hydrogen atoms and equivalent isotropic thermal parameters $(Å^2 \times 10^3)$ with e.s.d.'s in parentheses

$$U_{\rm eq} = \frac{1}{3} \sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot \mathbf{a}_j.$$

x	У	Z	U_{eq}
8831 (3)	3003 (13)	4086 (19)	30 (3)
9077 (3)	1192 (13)	3792 (19)	33 (3)
8730 (4)	22 (14)	2575 (22)	50 (4)
9573 (2)	1387 (9)	2852	38 (2)
9184 (4)	374 (13)	5782 (22)	38 (3)
9503 (4)	1553 (14)	7090 (19)	40 (3)
9601 (3)	734 (11)	8953 (15)	53 (2)
9181 (4)	3174 (13)	7530 (20)	42 (4)
9070 (4)	4222 (13)	5671 (22)	41 (4)
8679 (4)	5630 (15)	6190 (25)	62 (5)
9559 (4)	5080 (15)	4785 (23)	59 (4)
8455 (4)	3495 (16)	2966 (24)	54 (4)
8084 (5)	3992 (18)	1872 (25)	80 (6)
7528 (5)	3865 (23)	2350 (32)	102 (8)
7382 (5)	3204 (21)	4281 (35)	94 (7)
7198 (4)	4227 (21)	1158 (29)	180 (8)

Table 2. Interatomic distances (Å)

C2–C1	1.564 (14)	C5-C4	1.546 (14)
C6–C1	1.575 (15)	C6-C5	1.547 (17)
C7C1	1.302 (15)	C61–C6	1.537 (15)
C21–C2	1.532 (15)	C62–C6	1.559 (15)
O2–C2	1.447 (11)	C8–C7	1.281 (17)
C3C2	1.537 (15)	C9C8	1.483 (18)
C4–C3	1.530 (15)	C91–C9	1.476 (25)
O4C4	1.456 (13)	O9-C9	1.220 (18)

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^{*} The authors apologize for the lack of detail concerning the experimental conditions of the data collection. The data were collected by Fleischer in 1969 and the outputs containing the relevant details have become lost in the intervening years. An *hk*0 projection $R \sim 20\%$ was obtained in 1969 but no success was achieved in three dimensions. The data were sent by Fleischer to Tollin in 1969 as 641 observed reflections and an unsuccessful attempt to solve the structure using Patterson methods was made. The advent of *MITHRIL* was the key which unlocked the door.

parameters refined, R = 0.069, wR = 0.070, unit weights. Max. Δ/σ 0.01, max. difference map peak 0.22, min. $-0.24 \text{ e} \text{ Å}^{-3}$. In view of the high temperature factors for atoms C91 and O9 [largest values O9: U_{33} 0.247 Å², U_{22} 0.203 Å²; C91: U_{33} 0.146 Å²] the electron density in the vicinity was examined closely for indications of statistical disorder of these atoms. There were no such indications. Other programs used were XANADU (Roberts & Sheldrick, 1975) and PLUTO (Motherwell & Clegg, 1978). Scattering factors from International Tables for X-ray Crystallography (1974). No correction for secondary extinction.

Discussion. Atomic numbering is shown in Fig. 1. Atomic coordinates of the non-hydrogen atoms are given in Table 1,* with bond lengths in Table 2. There are two hydrogen bonds in the structure both involving O2 and O4. $O4\cdots O2(2-x, -y, \frac{1}{2}+z) 2.81$ (2) Å and $O2\cdots O4(x, y, -1+z) 2.74$ (2) Å. O9 is not involved in any close contacts. The cyclohexane ring has a chair conformation. Fig. 2 is a view of the unit cell showing hydrogen bonding.

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Fig. 2. View of unit cell along b showing hydrogen bonding as thin lines. Symmetry codes: (i) x, y, -1 + z; (ii) 2 - x, -y, $\frac{1}{2} + z$.

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Molecular Structure Analysis of Benzamide Neuroleptics and Analogs. XI. exo-2,3-Dihydro-N-(8-benzyl-8-azabicyclo[3.2.1]oct-3-yl)-1,3-benzodioxole-5-carboxamide and exo-2,3-Dihydro-N-(8-benzyl-8-azabicyclo[3.2.1]oct-3-yl)-1,4-benzodioxin-4-carboxamide

BY SONIA COLLIN, BERNADETTE NORBERG, GUY EVRARD AND FRANÇOIS DURANT

Laboratoire de Chimie Moléculaire Structurale, Facultés Universitaires Notre-Dame de la Paix, Rue de Bruxelles 61, B-5000 Namur, Belgium

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Abstract. (I) $C_{22}H_{24}N_2O_3$. $M_r = 364.4$, monoclinic, $P2_1/n$, a = 12.833 (5), b = 12.012 (6), c = 12.515 (5) Å, $\beta = 101.12$ (3)°, V = 1893.0 Å³, Z = 4, $D_x = 1.28 \text{ g cm}^{-3}$, $\lambda(Mo K\alpha) = 0.71069$ Å, $\mu = 0.49 \text{ cm}^{-1}$, F(000) = 776, R = 0.04 for 1720 reflections. (II) $C_{22}H_{25}N_2O_3^+$.Cl⁻.H₂O. $M_r = 418.9$, monoclinic, $P2_1/c$, a = 15.580 (1), b = 6.865 (1), c = 19.928 Å, $\beta = 104.85$ (1)°, V = 2060.2 Å³, Z = 4, $D_x = 1.35$ g cm⁻³, λ (Cu K α) = 1.54178 Å, $\mu = 17.94$ cm⁻¹, F(000) = 888, R = 0.04 for 1929 reflectons. (III) $C_{23}H_{26}N_2O_3$. $M_r = 378.5$, monoclinic, $P2_1/n$, a = 12.994 (3), b = 12.435 (3), c = 12.417 (4) Å, β

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^{*} Lists of structure factors, anisotropic thermal parameters, H-atom parameters and bond angles have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 43504 (8 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.